# Plant Peptide Hormone Phytosulfokine (PSK- $\alpha$ ): Synthesis of New Analogues and Their Biological Evaluation 

AGATA BAHYRYCZ, ${ }^{a}$ YOSHIKATSU MATSUBAYASHI, ${ }^{\text {b }}$ MARI OGAWA, ${ }^{b}$ YOUJI SAKAGAMI ${ }^{b}$ and DANUTA KONOPIŃSKA ${ }^{\text {a* }}$<br>a Faculty of Chemistry, University of Wrocław, 50-383 Wrocław, ul. F. Joliot-Curie 14, Poland<br>${ }^{\text {b }}$ Graduate School of Bio-Agricultural Sciences, Nagoya University, Chikusa, Nagoya 464-8601, Japan

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#### Abstract

Phytosulfokine- $\alpha$ ( $\mathrm{PSK}-\alpha$ ), a sulfated growth factor ( $\mathrm{H}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right.$ )-Ile- $\operatorname{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$ - $\mathrm{Thr}-\mathrm{Gln}-\mathrm{OH}$ ) universally found in both monocotyledons and dicotyledons, strongly promotes proliferation of plant cells in culture. In our studies on structure/activity relationship in PSK- $\alpha$ the synthesis of a series of analogues was performed: $\left[\mathrm{H}-\mathrm{d}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)^{1}\right]-(\mathbf{9})$, $\left[\mathrm{H}-\mathrm{Phe}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)^{1}\right]-(\mathbf{1 0})$, $\left[\mathrm{H}-\mathrm{d}-\mathrm{Phe}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)^{1}\right]-(\mathbf{1 1})$, $\left[\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)^{1}\right]-(\mathbf{1 2})$, $\left[\mathrm{H}-\mathrm{d}-\mathrm{Ph}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)^{1}\right]-(\mathbf{1 3}), \mathrm{H}-\mathrm{Phe}\left(4-\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)^{1}$ ]- (14), [H-d-Phe $\left(4-\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)^{1}$ ]- (15), [H-Phe $\left(4-\mathrm{NO}_{2}\right)^{1}$ ](16), [H-D-Phe $\left.\left(4-\mathrm{NO}_{2}\right)^{1}\right]-(\mathbf{1 7})$, $\left[\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{NO}_{2}\right)^{1}\right]-(\mathbf{1 8})$, $\left[\mathrm{H}-\mathrm{D}-\mathrm{Phg}\left(4-\mathrm{NO}_{2}\right)^{1}\right]-(\mathbf{1 9})$, $\left[\mathrm{H}-\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)^{1}\right]-(\mathbf{2 0})$, [H-$\left.\operatorname{Phg}\left(4-\mathrm{OSO}_{3} \mathrm{H}\right)^{1}\right]-(\mathbf{2 1})$, $\left[\mathrm{Phe}\left(4-\mathrm{NO}_{2}\right)^{3}\right]-(\mathbf{2 2})$, $\left[\mathrm{Phg}\left(4-\mathrm{NO}_{2}\right)^{3}\right]-(\mathbf{2 3})$, $\left[\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)^{3}\right]-(\mathbf{2 4})$, $\left[\mathrm{H}-\mathrm{Phe}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)^{1}\right.$, Phe $\left.\left(4-\mathrm{SO}_{3} \mathrm{H}\right)^{3}\right]-(\mathbf{2 5})\left[\mathrm{H}-\mathrm{Phe}\left(4-\mathrm{NO}_{2}\right)^{1}\right.$, $\left.\operatorname{Phe}\left(4-\mathrm{NO}_{2}\right)^{3}\right]-(\mathbf{2 6})$, $\left[\mathrm{H}-\mathrm{Ph} g\left(4-\mathrm{NO}_{2}\right)^{1}\right.$, $\left.\operatorname{Phg}\left(4-\mathrm{NO}_{2}\right)^{3}\right]-(\mathbf{2 7})$, $[\mathrm{H}-\mathrm{Hph}(4-$ $\left.\left.\mathrm{NO}_{2}\right)^{1}, \mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)^{3}\right]-(\mathbf{2 8})$ and $\left[\mathrm{Val}^{3}\right]-\mathrm{PSK}-\alpha$ (29). For modification of the PSK- $\alpha$ peptide chain the novel amino acids and their derivatives were synthesized, such as: $\mathrm{H}-\mathrm{L}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}(\mathbf{1}), \mathrm{H}-\mathrm{d}-\mathrm{Phg}(4-\mathrm{SO} 3 \mathrm{H})-\mathrm{OH}$ (2), Fmoc-Phg(4-SO $\left.3_{3} \mathrm{H}\right)-\mathrm{OH}(\mathbf{3})$, $\mathrm{Fmoc}-\mathrm{d}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}(4)$, $\mathrm{Boc}-\mathrm{Phg}\left(4-\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)$-OH (5), Boc-d-Phg(4$\left.\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)-\mathrm{OH}(6) \mathrm{Boc}-\mathrm{Phe}\left(4-\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)-\mathrm{OH}(7)$, and Boc-d-Phe(4-NHSO $\left.2 \mathrm{CH}_{3}\right)-\mathrm{OH}$ (8). Peptides were synthesized by a solid phase method according to the Fmoc procedure on a Wang-resin. Free peptides were released from the resin by $95 \% \mathrm{TFA}$ in the presence of EDT. All peptides were tested by competitive binding assay to the carrot membrane using ${ }^{3} \mathrm{H}$-labelled PSK according to the Matsubayashi et al. test. Copyright © 2004 European Peptide Society and John Wiley \& Sons, Ltd.


Keywords: plant peptide hormone; phytosulfokine (PSK- $\alpha$ )

## INTRODUCTION

The sulfated peptide phytosulfokine (PSK) is an intercellular signal that plays a key role in cellular de-differentiation and re-differentiation in plants [1]. Sulfated tyrosine residues are often found in secreted peptides in animals, but to date PSK is the only example of post-translational sulfation

[^0]of tyrosine residues in plants. Several paralogous genes encoding $\approx 80$-residue precursors of PSK have been identified in Arabidopsis. Each predicted protein has a probable secretion signal at the $N$-terminus and a single PSK sequence close to the $C$-terminus, similar to other peptide hormones generally synthesized as inactive higher molecular weight precursors which must undergo a variety of post-translational processing steps to yield the active peptides [2]. Studies using radiolabelled PSK have provided evidence for the existence of highaffinity binding sites for PSK in plant plasma membranes [2,3]. Recently, PSK receptor has been purified from membrane fractions and cloned carrot cells [4]. The cDNA encodes a typical LRR receptor
kinase that has 21 LRRs and a 36-residue island between the 17th and 18th LRRs.

The biological evaluation of unsulfated analogues of these peptides showed that these peptides required the sulfate ester for the expression of their biological activity [5,6]. Moreover, Matsubayashi et al. [7] observed that the unsulfated PSK- $\alpha$ analogue was dramatically less active. Based on the above results studies were performed on the structure/activity relationship in PSK- $\alpha$ and three series of multidirectional modified analogues obtained: (1) analogues modified in position 1 by different non-protein aromatic amino acid residues (9-21): H-d-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (9), H-Phe(4- $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Tyr $\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-GlnOH (10), H -d-Phe( $4-\mathrm{SO}_{3} \mathrm{H}$ )-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-GlnOH (11), $\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Tyr $\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-GlnOH (12), $\mathrm{H}-\mathrm{d}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)$-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-GlnOH (13), H -Phe( $4-\mathrm{NHSO}_{2} \mathrm{CH}_{3}$ )-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (14), H-d-Phe(4- $\left.\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)$-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$ -Thr-Gln-OH (15), H-Phe(4- $\mathrm{NO}_{2}$ )-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (16), H-d-Phe(4-NO2)-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (17), H-Phg(4-NO $\mathrm{N}_{2}$ )-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-GlnOH (18), H -d-Phg $\left(4-\mathrm{NO}_{2}\right)$-Ile-Tyr $\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln$\mathrm{OH}(\mathbf{1 9}), \mathrm{H}-\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)$-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (20), $\quad \mathrm{H}-\mathrm{Phg}\left(4-\mathrm{OSO}_{3} \mathrm{H}\right)$-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (21); (2) analogues modified in position 3 by $4-\mathrm{NO}_{2}$ Phe, Phg and Hph (22-24): $\mathrm{H}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Phe(4$\mathrm{NO}_{2}$ )-Thr-Gln-OH (22), $\mathrm{H}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Phg(4$\mathrm{NO}_{2}$ )-Thr-Gln-OH (23), $\quad \mathrm{H}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{Ile}-\mathrm{Hph}(4-$ $\mathrm{NO}_{2}$ )-Thr-Gln-OH (24); (3) analogues modified in both positions 1 and 3 by $4-\mathrm{SO}_{3} \mathrm{H}$ or 4 $\mathrm{NO}_{2}$ aromatic amino acids (25-28), such as: H-Phe( $4-\mathrm{SO}_{3} \mathrm{H}$ )-Ile-Phe (4-SO ${ }_{3} \mathrm{H}$ )-Thr-Gln-OH (25),

H -Phe $\left(4-\mathrm{NO}_{2}\right)$-Ile-Phe( $4-\mathrm{NO}_{2}$ )-Thr-Gln-OH (26), $\mathrm{H}-$ Phg (4- $\mathrm{NO}_{2}$ )-Ile-Phg $\left(4-\mathrm{NO}_{2}\right)$-Thr-Gln-OH (27), $\mathrm{H}-$ $\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)$-Ile- $\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)$-Thr-Gln-OH (28) and $\mathrm{H}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Val-Gln-OH (29).

In the first group of analogues the 4 -sulfatedTyr residue in position 1 was replaced by: (a) its d-isomer (9), (b) L - or d -Phe derivatives 4 -substituted by $-\mathrm{SO}_{3} \mathrm{H},-\mathrm{NO}_{2}$ or $-\mathrm{NHSO}_{2} \mathrm{CH}_{3}$ groups, (c) amino acids deprived of the methylene group between the $C^{\alpha}$ atom and the phenyl ring (4-substituted L - or d-phenylglycine derivatives) (10-19), or (d) amino acid containing two methylene groups between the $C^{\alpha}$ atom and the phenyl ring (4-nitro-l-homo-phenylalanine (20)). In analogues modified in position 3, the sulfonated Tyr residue was replaced by 4-nitro-derivatives of Phe, Phg and Hph (22-24). In the third series of analogues of PSK- $\alpha$ modified in both position 1 and 3, the $\operatorname{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$ residues were replaced by Phe $\left(4-\mathrm{SO}_{3} \mathrm{H}\right)$, $\operatorname{Phe}\left(4-\mathrm{NO}_{2}\right)$, $\mathrm{Phg}\left(4-\mathrm{NO}_{2}\right)$ and $\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)$ (25-28). Moreover, the synthesis was performed of PSK analogues modified in position 4 by Val, an amino acid with the isosteric side chain relative to Thr (analogue 29). For modification of the above peptides the novel amino acid derivatives were synthesized: $\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}$ (1), $\mathrm{H}-\mathrm{d}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}$ (2), Fmoc-Phg(4- $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-OH (3), Fmoc-d-Phg(4-SO $\left.3_{3} \mathrm{H}\right)$ $\mathrm{OH}(\mathbf{4}), \mathrm{Boc}-\mathrm{Phg}\left(4-\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)-\mathrm{OH}(5)$, Boc-d-Phg(4$\mathrm{NHSO}_{2} \mathrm{CH}_{3}$ )-OH (6) Boc-Phe(4-NHSO $\left.2 \mathrm{CH}_{3}\right)-\mathrm{OH}(\mathbf{7})$, and Boc-d-Phe(4-NHSO $2_{2} \mathrm{CH}_{3}$ )-OH (8) (Scheme 1 and Scheme 2).
Synthesis of these peptides was performed by the solid phase method according to standard procedure. Amino acid derivatives were assembled


Scheme 1 Synthesis of Fmoc-Phg(4-SO3 H)-OH.


Scheme 2 Synthesis of Boc-Phg(4-NHSO $\left.2_{2} \mathrm{CH}_{3}\right)-\mathrm{OH}$.


Scheme 3 Synthesis of $\left[\mathrm{H}-\mathrm{d}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)^{1}\right]$-PSK- $\alpha$.
on a Wang-resin, according to the Fmoc-procedure. Free peptides were released from the resin by $95 \%$ TFA in the presence of EDT (Scheme 3).

All peptides were tested according to the method of Matsubayashi et al. [4].

## MATERIAL AND METHODS

## Chemical Part

General procedures. Amino acid compositions were determined on an amino acid analyser Mikrotechna T339 (Czechoslovakia). The optical activity of the chiral compounds was measured with a Jasco DIP-1000 polarimeter (Jasco, Japan). The molecular weights of the peptides were determined using a Finigan Mat TSQ 700 (USA) mass spectrometer. The purity and homogeneity of all final products were checked by HPLC (Beckman Peptide Gold System) and TLC on silica gel
plates, amino acid analysis and molecular weight determinations. The purity of all peptides was about $100 \%$.
$N$-protected amino acid derivatives: FmocThr( $\left.\mathrm{Bu}^{t}\right)-\mathrm{OH}$, Fmoc-Ile-OH, Fmoc-Val-OH and Fmoc-Tyr-OH (Novabiochem) were used. Other amino acid derivatives, such as H - $\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)$ $\mathrm{OH}, \mathrm{H}-\mathrm{d}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}$, and $N$-Boc- or $N$ -Fmoc-derivatives $\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}$, $\mathrm{H}-\mathrm{Phg}(4-\mathrm{OH})-$ $\mathrm{OH}, \mathrm{H}-\mathrm{d}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}, \mathrm{H}-\mathrm{Phe}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}, \mathrm{H}-$ d-Phe(4- $\left.\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}, \quad \mathrm{H}-\mathrm{Phg}\left(4-\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)-\mathrm{OH}, \quad \mathrm{H}-$ d-Phg(4- $\left.\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)$ - OH , H-Phe $\left(4-\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)-\mathrm{OH}$, $\mathrm{H}-\mathrm{d}-\mathrm{Phe}\left(4-\mathrm{NHSO}_{2} \mathrm{CH}_{3}\right)-\mathrm{OH}, \mathrm{H}-\mathrm{Phe}\left(4-\mathrm{NO}_{2}\right)-\mathrm{OH}, \mathrm{H}-$ d-Phe $\left(4-\mathrm{NO}_{2}\right)-\mathrm{OH}, \quad \mathrm{H}-\mathrm{Phg}\left(4-\mathrm{NO}_{2}\right)-\mathrm{OH}, \quad \mathrm{H}-\mathrm{d}-\mathrm{Phg}(4-$ $\left.\mathrm{NO}_{2}\right)-\mathrm{OH}$, and $\mathrm{H}-\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)$-OH were synthesized in our laboratory, according to [8-13] (Table 1). All peptides (9-29) were obtained by the solid-phase method according to the Fmoc procedure. Amino acids were assembled on a Fmoc-Gln(Trt)-Wang resin (Novabiochem). $N$-terminal residues were introduced as $N$-Boc-derivatives, except peptides 9,
Table 1 Physicochemical Data of Phg and Phe Derivatives

| Compound | Yield <br> (\%) | $\begin{aligned} & \text { M.p } \\ & \left({ }^{\circ} \mathrm{C}\right) \end{aligned}$ | $[\alpha]_{D}^{20}$ | Analysis |  |  |  |  |  |  |  | Mw |  | $\mathrm{TLC}^{\text {a }} R_{\mathrm{f}}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | \%C |  | \%H |  | \%N |  | \%S |  | Calc. | Found | X | Y | Z |
|  |  |  |  | Calc. | Found | Calc. | Found | Calc. | Found | Calc. | Found |  |  |  |  |  |
| $\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)-\mathrm{OH}(\mathbf{1})$ | 82.2 | >250 | $+62.2{ }^{\text {b }}$ | 41.5 | 41.3 | 3.9 | 3.7 | 6.0 | 6.1 | 13.9 | 13.8 | 231.1 | 231.2 | 0.125 | 0.13 | 0.73 |
| $\mathrm{H}-\mathrm{d}$-Phg(4-SO ${ }_{3} \mathrm{H}$ )-OH (2) | 87.9 | >250 | $-62.2^{\text {b }}$ | 41.5 | 41.5 | 3.9 | 3.6 | 6.0 | 6.2 | 13.9 | 13.7 | 231.1 | 231.6 | 0.13 | 0.11 | 0.76 |
| Fmoc-Phg(4-SO3H)-OH (3) | 58.4 | 134-137 | $+3.1{ }^{\text {c }}$ | 61.1 | 61.3 | 4.0 | 3.9 | 3.1 | 3.1 | 7.1 | 7.1 | 452.4 | 452.2 | 0.51 | 0.56 | 0.84 |
| Fmoc-d-Phg(4-SO3 S )-OH (4) | 60.5 | 138-140 | $-3.1{ }^{\text {b }}$ | 61.1 | 61.1 | 4.0 | 3.9 | 3.1 | 3.0 | 7.1 | 7.1 | 452.4 | 453.4 | 0.52 | 0.58 | 0.82 |
| Boc-Phg(4-NHSO ${ }_{2} \mathrm{CH}_{3}$ )-OH (5) | 58.4 | 118-120 | $+97.5^{\text {d }}$ | 48.9 | 48.4 | 5.8 | 5.9 | 8.1 | 8.7 | 9.3 | 9.2 | 343.4 | 344.5 | 0.84 | 0.81 | 0.79 |
| Boc-d-Phg(4- $\mathrm{NHSO}_{2} \mathrm{CH}_{3}$ )-OH (6) | 80.1 | 119-121 | $-97.5^{\text {d }}$ | 48.9 | 48.3 | 5.8 | 5.4 | 8.1 | 8.5 | 9.3 | 9.2 | 343.4 | 344.6 | 0.82 | 0.80 | 0.76 |
| Boc-Phe(4-NHSO ${ }_{2} \mathrm{CH}_{3}$ )-OH (7) | 60.4 | 130-133 | $+38.4{ }^{\text {d }}$ | 50.3 | 50.5 | 6.1 | 5.9 | 7.8 | 7.7 | 8.9 | 9.0 | 358.1 | 357.3 | 0.24 | 0.57 | 0.43 |
| Boc-d-Phe(4-NHSO $2_{2} \mathrm{CH}_{3}$ )-OH (8) | 65.1 | 129-132 | $-38.1{ }^{\text {d }}$ | 50.3 | 50.2 | 6.1 | 6.0 | 7.8 | 7.9 | 8.9 | 8.8 | 358.1 | 358.7 | 0.27 | 0.53 | 0.41 |

[^1]12 and 13. For these peptides $N$-Fmoc derivatives were used. The $C$-terminal Gln residue was bound to the resin as Fmoc-Gln(Trt) (Novabiochem). HBTU in the presence of HOBt and $N$-ethylmorpholine were used as coupling reagents. During the synthesis the Tyr residue was used as Fmoc-Tyr-OH. The $N^{\alpha}$-Fmoc group was removed with $20 \%$ piperidine in $N, N-$ dimethylformamide (DMF) according to standard methods. The partially protected peptide-resin in DMF-pyridine ( $4: 1$ ) was sulfated by DMF- $\mathrm{SO}_{3}$ complex. The sulfated peptide-resin was cleaved in $95 \%$ trifluoroacetic acid (TFA) in the presence of ethanedithiol (EDT)

All peptides (9-29) were purified by semipreparative HPLC on an Alltech Econsil C $\mathrm{C}_{18}$, $10 \mu \mathrm{~m}$ column (ODS $22 \times 250 \mathrm{~mm}$ ), linear gradient $23 \%-39 \% \mathrm{~S} 2$ for 15 min , flow rate $7 \mathrm{ml} / \mathrm{min}$, determined at 223 nm
Analytical RP-HPLC was conducted on a Beckman Peptide Gold System chromatograph with $\mathrm{C}_{18}, 5 \mu \mathrm{~m}$ Beckman column (ODS $4.6 \times 250 \mathrm{~mm}$ ), ultrasphere plus $4.6 \times 4.5 \mathrm{~mm}$ precolumn. Solvent systems: Sl$0.1 \%$ aqueous TFA, S2-80\% acetonitrile; linear gradient from $0-100 \%$ of S 2 for 60 min , flow rate $1.0 \mathrm{ml} / \mathrm{min}$, determined at 223 nm . An isocratic system ( $18 \%$ acetonitrile) was also applied to check the purity.

Purity and homogeneity of the free peptides were established by amino acid analysis and determination of molecular weights and optical activity. The physico-chemical data PSK- $\alpha$ analogues are summarized in Table 2.

4-Sulfo-L-phenylglycine. H -Phg(4- $\mathrm{SO}_{3} \mathrm{H}$ )-OH (1). The title compound was synthesized according to [8]. Phg ( $15.1 \mathrm{~g}, 0.1 \mathrm{~mol}$ ) was dissolved in concentrated chlorosulfonic acid ( 45 ml ) at $15^{\circ} \mathrm{C}$. The solution was heated to $36^{\circ} \mathrm{C}$, mixed for 30 h , and finally dropped to ice. The white product was crystallized from isopropanol: water ( $1: 1$ ). 19 g of the product was obtained. Physico-chemical data are presented in Table 1.

## 4-Sulfo-D-phenylglycine. H-D-Phg(4-SO ${ }_{3} \mathrm{H}$ )-OH (2).

 The title compound was obtained in the same manner as compound $\mathbf{1}$ from $15.1 \mathrm{~g}(0.1 \mathrm{~mol})$ of d-phenylglycine. Crystallization from isopropanol: water gave 20.3 g of the product. Physico-chemical data are presented in Table 1.
## $\mathrm{N}^{\alpha}$-(9-Fluorenylmethoxycarbonyl)-4-sulfo-L

phenylglycine. Fmoc-Phg(4-SO3 H)-OH (3). To introduce the Fmoc group sulfo-L-phenylglycine ( $19.5 \mathrm{~g}, 0.08 \mathrm{~mol}$ ), Fmoc-OSu ( $27.0 \mathrm{~g}, 0.08 \mathrm{~mol}$ ) and
$\mathrm{Et}_{3} \mathrm{~N}(11 \mathrm{ml}, 0.08 \mathrm{~mol})$ were suspended in a mixture of $\mathrm{H}_{2} \mathrm{O}(100 \mathrm{ml})$ and THF ( 50 ml ). This heterogeneous mixture was stirred at $25^{\circ} \mathrm{C}$ for 2 h , while the pH was adjusted to $8.5-9.0$ with additional $\mathrm{Et}_{3} \mathrm{~N}$ until the pH was constant. The resultant homogeneous solution was concentrated in vacuo, and both EtOAc ( 100 ml ) and $\mathrm{H}_{2} \mathrm{O}(100 \mathrm{ml})$ were added. The mixture was acidified to pH 2.0 with 1 m HCl and the organic phase was washed with $5 \%$ citric acid ( $2 \times 40 \mathrm{ml}$ ), $\mathrm{H}_{2} \mathrm{O}(2 \times 40 \mathrm{ml})$, saturated aqueous $\mathrm{NaCl}(2 \times 40 \mathrm{ml})$ and dried over $\mathrm{MgSO}_{4}$. After concentration in vacuo the product was crystallized from ethyl acetate: pentane. 21.7 g of product was obtained (Scheme 1).

## $\mathrm{N}^{\alpha}$-(9-Fluorenylmethoxycarbonyl)-4-sulfo-Dphenylglycine. Fmoc-D-Phg( $\mathrm{SO}_{3} \mathrm{H}$ )-OH (4). 19.7 g

 ( 0.085 mol ) of compound 2 was reacted with 29 g ( 0.085 mol ) of $\mathrm{Fmoc}-\mathrm{OSu}$ in the same manner as in the case of 3. 22.5 g of the product was obtained (Table 1).
## $\mathrm{N}^{\alpha}$-(tert-Butoxycarbonyl)-4-aminosulfomethyl-Lphenylglycine. Boc-Phg(4-NHSO2 $\mathrm{CH}_{3}$ )-OH (5).

Boc-Phg $\left(4-\mathrm{NO}_{2}\right)$-OH [10] was dissolved in 40 ml of methanol and the solution was hydrogenated in the presence of $10 \% \mathrm{Pd} / \mathrm{C}(0.1 \mathrm{~g})$ for 48 h . Boc-Phg(4-$\left.\mathrm{NH}_{2}\right)-\mathrm{OH}(16.5 \mathrm{~g}, 0.062 \mathrm{~mol})$ was dissolved in 1 N $\mathrm{NaOH}(50 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$, and 10 ml of $\mathrm{CH}_{3} \mathrm{SO}_{2} \mathrm{Cl}$ (in 100 ml acetone) was added. The reaction was carried out for 2 h . Then the solvent was evaporated in vacuo, the aqueous phase was acidified to $\mathrm{pH} \sim 3$ with 1 m HCl and extracted with ethyl acetate. The organic phase was washed three times with water, dried over anhydrous $\mathrm{MgSO}_{4}$ and evaporated in vacuo. After crystallization from ethyl acetate: hexane 12.5 g of the product was obtained (Scheme 2).

## $\mathrm{N}^{\alpha}$-(tert-Butoxycarbonyl)-4-aminosulfomethyl-Dphenylglycine. Boc-D-Phg(4-NHSO2 $\mathrm{CH}_{3}$ )-OH (6).

 The title compound was obtained in the same manner as compound $\mathbf{5}$. The reaction was carried out as described above. 15.6 g of the product was obtained (Table 1).
## $\mathrm{N}^{\alpha}$-(tert-Butoxycarbonyl)-4-aminosulfomethyl-Lphenylalanine. Boc-Phe(4-NHSO2 $\mathrm{CH}_{3}$ )-OH (7).

Boc-Phe ( $4-\mathrm{NO}_{2}$ )-OH [13] ( $22 \mathrm{~g}, 0.08 \mathrm{~mol}$ ) was dissolved in 40 ml of methanol and the solution was hydrogenated in the presence of $10 \% \mathrm{Pd} / \mathrm{C}(0.1 \mathrm{~g})$ for $48 \mathrm{~h} .15 .0 \mathrm{~g}(0.054 \mathrm{~mol})$ of Boc-Phe $\left(4-\mathrm{NH}_{2}\right)-\mathrm{OH}$ was obtained. The product was dissolved in 1 N NaOH ( 50 ml ) at $0^{\circ} \mathrm{C}$, and then 8.7 ml of $\mathrm{CH}_{3} \mathrm{SO}_{2} \mathrm{Cl}$ (in
Table 2 Physicochemical Data of PSK Analogues Modified in Position 1, 3 and 4 of the Peptide Chain

| Peptide | Yield <br> (\%) | $\begin{gathered} {[\alpha]_{\mathrm{D}}^{20}} \\ 1.0 \% \mathrm{NH}_{4} \mathrm{OH} \end{gathered}$ |  | $\begin{gathered} \mathrm{Rt}^{\mathrm{a}} \\ \text { (HPLC) } \end{gathered}$ | Amino acid analysis | Mw |  | TLC ${ }^{\text {b }} R_{\text {f }}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | Calc. |  | Found | X | Y | Z |
| $\mathrm{H}-\mathrm{d}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (9) | 56 | -26.3 | $\mathrm{c}=0.25$ |  | 18.16 | Tyr 1.8 Ile 1.2 Thr 1.0 Gln 1.0 | 846.2 | 845.6 | 0.22 | 0.67 | 0.49 |
| H -Phe( $\mathrm{SO}_{3} \mathrm{H}$ )-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (10) | 71 | -2.4 | $\mathrm{c}=0.5$ | 7.17 | Ile 0.8 Tyr 1.0 Thr 1.2 Gln 1.0 | 830.2 | 829.4 | 0.10 | 0.64 | 0.40 |
| H-d-Phe(4-SO3 H)-Ile-Tyr( $\mathrm{SO}_{3} \mathrm{H}$ )-Thr-Gln-OH (1 1) | 58 | -22.5 | $\mathrm{c}=0.5$ | 14.05 | Ile 0.8 Tyr 1.1 Thr 0.9 Gln 0.9 | 830.2 | 829.7 | 0.13 | 0.44 | 0.48 |
| $\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (12) | 61 | -1.1 | $\mathrm{c}=1.0$ | 11.85 | Ile 1.09 Tyr 0.9 Thr 0.8 Gln 1.0 | 816.2 | 815.1 | 0.12 | 0.63 | 0.42 |
| $\mathrm{H}-\mathrm{d}-\mathrm{Phg}\left(4-\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (13) | 60 | -1.1 | $\mathrm{c}=1.0$ | 12.11 | Ile 1.2 Tyr 0.9 Thr 1.0 Gln 0.98 | 816.2 | 815.9 | 0.11 | 0.62 | 0.41 |
| H-Phe(4-NHSO2 $\mathrm{CH}_{3}$ )-Ile-Tyr( $\mathrm{SO}_{3} \mathrm{H}$ )-Thr-Gln-OH (14) | 48 | -14.7 | $\mathrm{c}=0.5$ | 17.29 | Ile 1.05 Tyr 095 Thr 0.98 Gln 0.9 | 843.4 | 842.9 | 0.10 | 0.44 | 0.57 |
| H-d-Phe(4-NHSO ${ }_{2} \mathrm{CH}_{3}$ )-Ile-Tyr( $\mathrm{SO}_{3} \mathrm{H}$ )-Thr-Gln-OH (15) | 45 | -5.8 | $\mathrm{c}=0.5$ | 19.305 | Ile 1.1 Tyr 1.0 Thr 1.0 Gln 0.99 | 843.4 | 843.2 | 0.09 | 0.60 | 0.53 |
| H -Phe(4- $\mathrm{NO}_{2}$ )-Ile-Tyr $\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (16) | 50 | -24.7 | $\mathrm{c}=0.5$ | 20.62 | Ile 1.06 Tyr 1.1 Thr 0.99 Gln 0.98 | 795.2 | 794.7 | 0.15 | 5.54 | 0.55 |
| H-d-Phe( $4-\mathrm{NO}_{2}$ )-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (17) | 48 | -37.4 | $\mathrm{c}=0.5$ | 18.94 | Ile 0.96 Tyr 1.0 Thr 0.97 Gln 1.0 | 795.2 | 795.0 | 0.16 | 0.66 | 0.69 |
| $\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{NO}_{2}\right)$-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (18) | 54 | -16.4 | $\mathrm{c}=0.5$ | 22.36 | Ile 1.2 Tyr 1.11 Thr 0.9 Gln 1.0 | 781.2 | 780.6 | 0.16 | 0.68 | 0.70 |
| $\mathrm{H}-\mathrm{d}-\mathrm{Phg}\left(4-\mathrm{NO}_{2}\right)$-Ile-Tyr( $\left.\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (19) | 46 | -5.1 | $\mathrm{c}=0.5$ | 18.82 | Ile 1.0 Tyr 0.9 Thr 0.98 Gln 1.0 | 781.2 | 780.7 | 0.11 | 0.61 | 0.54 |
| $\mathrm{H}-\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)$-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (20) | 47 | -9.4 | $\mathrm{c}=0.5$ | 18.10 | Ile 1.2 Tyr 1.0 Thr 0.9 Gln 0.9 | 809.2 | 808.9 | 0.10 | 0.64 | 0.46 |
| $\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{OSO}_{3} \mathrm{H}\right)$-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Thr-Gln-OH (21) | 56 | -16.1 | $\mathrm{c}=0.5$ | 13.81 | Tyr 0.96 Ile 1.16 Thr 1.0 Gln 0.9 | 832.2 | 831.9 | 0.12 | 0.48 | 0.41 |
| $\mathrm{H}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Phe(4-NO2)-Thr-Gln-OH (22) | 49 | -1.8 | $\mathrm{c}=1.0$ | 19.30 | Ile 1.0 Tyr 0.9 Thr 0.9 Gln 0.98 | 795.2 | 796.0 | 0.11 | 0.51 | 0.43 |
| $\mathrm{H}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Phg(4-NO2)-Thr-Gln-OH (23) | 51 | -16.8 | $\mathrm{c}=0.5$ | 20.57 | Tyr 1.0 Ile 1.2 Thr 1.0 Gln 0.9 | 781.2 | 782.0 | 0.10 | 0.61 | 0.22 |
| $\mathrm{H}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Hph(4-NO2)-Thr-Gln-OH (24) | 48 | -15.8 | $\mathrm{c}=0.5$ | 18.82 | Tyr 1.1 Ile 1.0 Thr 0.9 Gln 0.88 | 809.2 | 810.1 | 0.14 | 0.67 | 0.57 |
| H -Phe $\left(4-\mathrm{SO}_{3} \mathrm{H}\right)$-Ile-Phe(4-SO $\left.3{ }_{3} \mathrm{H}\right)$-Thr-Gln-OH (25) | 61 | -24.0 | $\mathrm{c}=1.0$ | 10.76 | Ile 0.99 Tyr 1.0 Thr 1.0 Gln 0.9 | 814.2 | 813.3 | 0.10 | 0.36 | 0.22 |
| H-Phe(4-NO2)-Ile-Phe(4-NO2)-Thr-Gln-OH (26) | 90 | -3.9 | $\mathrm{c}=1.0$ | 23.52 | Ile 1.0 Thr 0.9 Gln 0.98 | 744.2 | 742.9 | 0.31 | 0.79 | 0.69 |
| $\mathrm{H}-\mathrm{Phg}\left(4-\mathrm{NO}_{2}\right)$-Ile-Phg(4-NO2)-Thr-Gln-OH (27) | 87 | -12.9 | $\mathrm{c}=0.5$ | 20.72 | Ile 1.2 Thr 1.0 Gln 1.0 | 716.2 | 717.0 | 0.30 | 0.68 | 0.68 |
| $\mathrm{H}-\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)$-Ile- $\mathrm{Hph}\left(4-\mathrm{NO}_{2}\right)$-Thr-Gln-OH (28) | 89 | -11.9 | $\mathrm{c}=0.5$ | 25.42 | Ile 1.0 Thr 0.9 Gln 0.98 | 772.2 | 773.1 | 0.26 | 0.72 | 0.66 |
| $\mathrm{H}-\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Ile- $\mathrm{Tyr}\left(\mathrm{SO}_{3} \mathrm{H}\right)$-Val-Gln-OH (29) | 48 | -11.0 | $\mathrm{c}=0.5$ | 15.95 | Tyr 1.98 Ile 1.02 Thr 0.99 Val 1.01 | 844.2 | 845.4 | 0.09 | 0.53 | 0.58 |

[^2]100 ml acetone) was added. The reaction was carried out for 2 h . Then the solvent was evaporated in vacuo, the aqueous phase was acidified to pH $\sim 3$ with 1 m HCl and extracted with ethyl acetate. The organic phase was washed three times with water, dried over anhydrous $\mathrm{MgSO}_{4}$ and evaporated in vacuo. After crystallization from ethyl acetate: hexane 11.6 g of the product was obtained.

## $\mathrm{N}^{\alpha}$-(tert-Butoxycarbonyl)-4-aminosulfomethyl-Dphenylalanine. Boc-D-Phe(4-NHSO2 $\mathrm{CH}_{3}$ )-OH (8).

The title compound was obtained in the same manner as compound 7. The reaction was carried out as described above. 12.5 g of the product was obtained (Table 1).

H-D-Tyr( $\mathrm{SO}_{3} \mathrm{H}$ )-Ile-Tyr( $\mathrm{SO}_{3} \mathrm{H}$ )-Thr-GIn-OH (9). The peptide was obtained by a stepwise elongation of the peptide chain by the method outlined above. 0.5 g of the Fmoc-Gln(Trt)-resin (substitution level $0.56 \mathrm{mmol} / \mathrm{g}$ ) was suspended in $20 \%$ solution of piperidine in DMF. The mixture was stirred for 20 min at room temperature. Then it was filtered and washed with DMF ( $5 \times 2 \mathrm{~min}$ ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( $5 \times 2 \mathrm{~min}$ ). The next amino acid, $\mathrm{Fmoc}-\mathrm{Thr}\left(\mathrm{Bu}^{t}\right)$ $\mathrm{OH}(0.33 \mathrm{~g}, 0.84 \mathrm{mmol})$, was dissolved in DMF and coupled to the resin in the presence of 1 equivalent of HBTU/HOBt and 2 equivalents of NEM ( $N$ ethylmorpholine) ( $198 \mu \mathrm{l}$ ) for 2 h . The end of the reaction was determined by the Kaiser test. Other Fmoc-amino acid derivatives: Fmoc-Tyr-OH, Fmoc-Ile-OH and Fmoc-d-Tyr-OH, were connected to the resin in the same way. Then the partially protected peptide-resin was sulfated by DMF-SO $3_{3}(2.6 \mathrm{~g}$, 30 equiv.) in DMF-pyridine ( $4: 1,8 \mathrm{ml}$ ) at room temperature for 16 h . The sulfated pentapeptideresin was collected by filtration, washed with water and dried overnight over KOH under reduced pressure. The $N^{\alpha}$-Fmoc group was subsequently removed with $20 \%$ piperidine in DMF. The free peptide was obtained by deprotection with 4.75 ml of TFA in the presence of 0.125 ml of ethanedithiol and 0.125 ml of water at room temperature according to standard procedure. Then the peptide was purified by preparative HPLC. The main fractions were pooled and lyophilized. The data are presented in Table 2.

Peptides 12-13 were obtained and purified in the same manner as peptide 9 (Table 2). Peptides 10-11 and 14-29 were obtained and purified in the same manner as peptide 9 except that the $N$ terminal amino acids were successively introduced as Boc amino acids. Their data are presented in Table 2.

## Biological Part

Competition binding of PSK analogues to PSK receptors in carrot microsome fractions was conducted in the presence of 3.2 nm of $\left[{ }^{3} \mathrm{H}\right]$ PSK and varying concentrations of the PSK analogues as previously described [4]. Error bars indicate $\pm$ SE from three independent experiments.

## RESULTS, DISCUSSION AND CONCLUSION

Among the analogues modified in position 1 of the peptide chain, only [Phe $\left.\left(4-\mathrm{NO}_{2}\right)^{1}\right]$ - (16) and [Phg(4-$\left.\left.\mathrm{NO}_{2}\right)^{1}\right]$-PSK- $\alpha$ (18) showed $10 \%$ binding activity compared with that of the native peptide, whereas [DPhg $\left.\left(4-\mathrm{NO}_{2}\right)\right]^{1}-\mathrm{PSK} \alpha(\mathbf{1 9})$ showed $1 \%$ binding activity. Other peptides were practically inactive. Basing on the preliminary biological activities obtained here it is difficult to discuss the structure/function relationship. Exchange of the oxygen atom at position $4^{\prime}$ of the aromatic ring of the $N$-terminal amino acid residue for the sulfur atom (analogues 10-13) or for the -NH system (peptides 14 and 15), as well as introducing the amino acid with configuration D (9) lead to derivatives with no biological effect in the carrot membrane competitive binding test. However, replacing of the $N$-terminal residue by aromatic amino acids, such as Phe(4$\mathrm{NO}_{2}$ ) (peptide 16), Phg $\left(4-\mathrm{NO}_{2}\right)$ (18) and $\mathrm{d}-\mathrm{Phg}(4-$ $\mathrm{NO}_{2}$ ) (19) lead to analogues with weak biological effects. These results point out that the presence of the $-\mathrm{OSO}_{3} \mathrm{H}$ system at the aromatic ring in position 1 of the PSK- $\alpha$ peptide chain or other substituent with the electro-acceptor character (such as the nitro group in the aromatic ring) plays an important role in creation of biological properties of PSK- $\alpha$. This problem will be a subject of further studies.

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[^0]:    *Correspondence to: Professor Danuta Konopińska, Faculty of Chemistry, University of Wrocław, 50-383 Wrocław, ul. F. JoliotCurie 14, Poland; e-mail: dk@wchuwr.chem.uni.wroc.pl
    The symbols of the amino acids, peptides and their derivatives are in accordance with the Recommendation of the IUPAC-IUB Joint Commission on Biochemical Nomenclature (1984) [Eur. J. Biochem. 1984; 138: 9-37].
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[^1]:    ${ }^{\text {a }}$ TLC on silica gel plates (Merck), eluents: $\mathrm{X}=n$-butanol:Ac-OH:water ( $4: 1: 5$ ), $\mathrm{Y}=n$-butanol:pyridine:Ac-OH:water ( $30: 20: 6: 24$ ), $Z=n$-butanol: AcOH : water ( $4: 1: 1$ ), ${ }^{\mathrm{b}} \mathrm{c}=1 \%$ in water, ${ }^{\mathrm{c}} \mathrm{c}=1 \%$ in DMF, ${ }^{\mathrm{d}} \mathrm{c}=1 \%$ in $\mathrm{CH}_{3} \mathrm{OH}$.

[^2]:    ${ }^{\text {a }}$ HPLC on Ultrasphere ODS columns (Beckman) $4.5 \times 250 \mathrm{~mm}$; solvent system: $\mathrm{S} 1-0.1 \%$ aqueous TFA, S2-80\% acetonitrile in water; linear gradient: $0-100 \%$
    ${ }^{\mathrm{b}} \mathrm{TLC}$ on silica gel plates (Merck), eluents: $\mathrm{X}=n$-butanol:Ac-OH: water ( $4: 1: 5$ ), $\mathrm{Y}=n$-butanol:pyridine:Ac-OH(30:20:6:24) $\mathrm{Z}=n$-butanol: Ac-OH:ethyl acetate: water $(1: 1: 1: 1)$.

